# Effects of CO<sub>2</sub> enrichment on plant-soil relationships of *Lepidium latifolium*

# Robert R. Blank<sup>1,3</sup> & Justin D. Derner<sup>2</sup>

<sup>1</sup>Soil Scientist U.S. Dept. of Agriculture, Agricultural Research Service, Exotic and Invasive Weed Research Unit, 920 Valley Road, Reno, NV, U.S.A. <sup>2</sup>Rangeland Scientist U.S. Dept. of Agriculture, Agricultural Research Service, High Plains Grasslands Research Station, Cheyenne, WY, U.S.A. <sup>3</sup>Corresponding author\*

Received 1 July 2003. Accepted in revised form 29 October 2003

Key words: carbon dioxide, competition, invasive species, Lepidium latifolium, plant-soil relationships, soil enzymes

#### **Abstract**

The exotic crucifer Lepidium latifolium L. (perennial pepperweed) is invading wetland and riparian habitats throughout the western United States. Based on previous field studies, our working hypothesis proposed that L. latifolium elevates soil nutrient acquisition ability in response to CO<sub>2</sub> enrichment. Replicates of L. latifolium were grown in a high fertility and low fertility soil (along with unplanted controls) in a glasshouse at ambient and elevated CO<sub>2</sub> concentrations (360 and 699  $\mu$ mol mol<sup>-1</sup>, respectively). Plants were harvested after 81 days and numerous plant and soil attributes measured. Above-ground plant mass was influenced by a significant CO<sub>2</sub> treatment  $\times$  soil interaction (P < 0.001) with CO<sub>2</sub> enrichment inducing a greater proportional increase in mass for the low fertility soil. Root concentrations of citrate, malate, and ortho-phosphate and enzyme activities of amidase and asparaginase did not differ between the CO<sub>2</sub> treatments across soils. Above-ground tissue concentrations of N, S, P, Mg, K, Fe, and Zn consistently decreased for both soils with CO<sub>2</sub> enrichment, corresponding with higher biomass per unit nutrient. Plants grown in the low fertility soil had higher concentrations of N, S, P, Ca, and Mg in above-ground tissue than plants grown in the high fertility soil. Carbon dioxide enrichment decreased tissue N:S ratios by > 20% and increased, though not significant, tissue C:N ratio by 38% in high fertility soil and by 51% in low fertility soil. For most soil attributes measured, there was a main effect or interaction with soil fertility level. Soil attributes differed between soil fertility levels and, with the exception of SO<sub>4</sub><sup>2-</sup>, were not influenced by the presence of L. latifolium. Soil attributes increased by CO<sub>2</sub> enrichment included acetate extractable Mg<sup>2+</sup> (high fertility soil only), net 30 day N mineralization potential (unplanted control soils only), available N (high fertility soil), bicarbonate extractable P, soil-solution  $SO_4^{2-}$  (L. latifolium planted pots only), and soil-solution Mg<sup>2+</sup> (high fertility control soil only). Collectively, these data tangentially support our working hypothesis that CO<sub>2</sub> enrichment increases nutrient availability. That availability of some nutrients increases without plant growth (control soils), however, suggests an interaction of elevated CO<sub>2</sub> with soil microflora.

# Introduction

The exotic crucifer *Lepidium latifolium* L. (perennial pepperweed), a native of southeastern Europe and Asia, has widely invaded wetland and riparian habitats throughout the western United States (Young et al., 1995). These C<sub>3</sub> plants are clonal and have extensive

underground, budding rootstocks that radiate in all directions from newly established plants. In 2 seasons, a single established plant becomes a small population that can be several meters in diameter. In as few as 5 years, infestations can be near monospecific with stem densities approaching  $150 \, \mathrm{m}^{-2}$  (Blank, 2002).

One potential explanation for the rapid invasion by *L. latifolium* is that the plant 'engineers' the soil to

<sup>\*</sup>FAX No: 775-784-1712. E-mail: blank@unr.nevada.edu

favor its own invasiveness (Jones et al., 1994). Previous field research demonstrated greater soil enzyme activity of several amidohydrolases in soils invaded by *L. latifolium* than in non-invaded soil (Blank, 2002). Moreover, amidohydrolase activities were significantly related to the KCl extractable N pool which suggests that *L. latifolium* had enhanced N availability. In that same study, roots of *L. latifolium* contained high levels of citrate and malate acids, which if exuded into the soil, may increase P availability (Hoffland et al., 1992). *L. latifolium* also increases Ca<sup>2+</sup> concentration in the soil solution, relative to non-invaded areas (Blank and Young, 2002).

Many invasive species capitalize on various elements of global change to become more successful (Dukes and Mooney, 1999). Mechanisms by which these invasive plants increase at the expense of existing plant species are poorly understood to date, but competitive interactions among species, which may be altered by CO<sub>2</sub> enrichment, are likely involved (Bazzaz and Garbutt, 1988; Derner et al., 2002; Marks and Strain, 1989; Ziska, 2001). Our construct visualizes increasing concentration of atmospheric carbon dioxide as currency, which the plant can spend in innumerable ways, predicated on genetics, positive and negative feedback loops, and internal and external stimuli. For example, insect herbivory on a plant leaf is an external stimuli which may cause a plant to increase production of secondary compounds to deter herbivory (Bazin et al., 2002). Another example is the finding that Lupinus albus exposed to elevated CO2 allocates more C to the production of proteoid roots which enhances greater P uptake (Campbell and Sage, 2002). We suggest that anthropogenic increases in atmospheric CO<sub>2</sub> have augmented the competitiveness of L. latifolium, and concomitantly its invasiveness, by improving plant-induced soil availability of N, P and Ca. We hypothesize that CO<sub>2</sub> enrichment 1) increases soil or plant root amidohydrolase and phosphatase activities (change in N or P availability via mineralization of organic matter), 2) increases citrate and malate in roots (change in soil P availability via root exudation), 3) increases pools of available soil Ca, and 4) alters plant-soil relationships of *L. latifolium*.

#### Materials and methods

Hypothesis testing took place in four glasshouses, two ambient and two elevated at the USDA-ARS Temple, TX ( $31^{\circ}05'$  N,  $97^{\circ}20'$  W) research facility. The CO<sub>2</sub>

concentration of air in each bay was measured at 4-min intervals with a model LI-6262 infrared gas analyzer (Li-Cor, Inc., Lincoln, NE, USA). The CO<sub>2</sub> readings were corrected for atmospheric pressure measured with a model DPI 260 pressure indicator (Druck, Inc., New Fairfield, CT, USA). The infrared analyzer was calibrated daily against four CO<sub>2</sub> gas standards and monthly against a LI-610 dewpoint generator (Li-Cor, Inc., Lincoln, NE, USA). Air temperature, manually set at 25 °C, was measured in the center of each bay with fine-wire (25- $\mu$ m diameter) thermocouples. Pure CO<sub>2</sub> gas was injected into the appropriate bays as required to maintain the elevated CO2 concentration. The CO<sub>2</sub> concentration of air in the ambient and elevated CO2 treatments averaged 360 and 699  $(\mu \text{mol mol}^{-1})$ , respectively. Photosynthetic photon flux density (PPFD) was measured on the glasshouse roof with a LI-190SB point quantum sensor (Li-Cor, Inc., Lincoln, NE, USA) and within the bays with 1-m long, LI-191SA, line quantum sensors (Li-Cor, Inc., Lincoln, NE, USA) mounted about plant height. On average, the daily integral of PPFD inside the bays was 70% of that measured above the glasshouse.

Two soil types were used: a high fertility substrate from the surface horizon of the Houston black series, a fine, smectitic, thermic Udic Haplustert, and a low fertility substrate from the surface horizon of the Pedernales series, a fine, mixed, thermic typic Palesustalf (Table 1). Although these soils are not present in areas where L. latifolium is invading, they do represent the range of soil fertility levels being invaded. Pots were filled to similar volumes with either 11.5 kg of the high fertility soil or 16 kg of the low fertility soil (different weights are because of bulk density differences of soils). The experiment was begun on 18 March, 2001 by planting 3 fresh root cuttings less than 5 cm in length of L. latifolium in 6 replicate pots  $\times$  2 soil types  $\times$  2 CO<sub>2</sub> treatments  $\times$  2 bays per CO<sub>2</sub> treatment = 48 total. We also filled pots with 3 replicates of unplanted control  $\times$  2 soil types  $\times$  2 CO<sub>2</sub> treatments  $\times$  2 bays per  $CO_2$  treatment = 24 total. A plastic barrier plate with a 3 cm rim was placed beneath each pot to prevent leaching. Pots were watered daily with deionzied water, but never to the extent that the barrier plates overflowed. After one root section sprouted in each pot, the other 2 sections were removed. Plants were harvested on 11 June, 2001, 81 days post-emergence. For each plant, leaves were removed at the petiole and leaf area quantified with a commercial leaf scanner (LI-3000A, Li-Cor, Inc. Lincoln, NE, USA). Root systems were cleansed of adhering soil by water and

Table 1. Selected initial soil attributes of the two soils<sup>a</sup>

| Attribute   | High fertility soil | Low fertility soil |
|---|---------------------|--------------------|
| Texture   | Clay                | Fine sandy loam    |
| KCl N (mmol kg <sup>-1</sup> )                              | 1.06(0.33)          | 0.14(0.03)         |
| Phosphatase ( $\mu$ mol g <sup>-1</sup> hr <sup>-1</sup> )  | 3.30(0.15)          | 0.27(0.04)         |
| Glutaminase ( $\mu$ mol g <sup>-1</sup> hr <sup>-1</sup> )  | 18.9(1.8)           | 0                  |
| Asparaginase ( $\mu$ mol g <sup>-1</sup> hr <sup>-1</sup> ) | 2.86(0.32)          | 0.06(0.02)         |
| Amidase ( $\mu$ mol g <sup>-1</sup> hr <sup>-1</sup> )      | 12.32(1.52)         | 0.44(0.04)         |
| Urease ( $\mu$ mol g <sup>-1</sup> hr <sup>-1</sup> )       | 4.56(0.79)          | 0.34(0.22)         |
| CaCl <sub>2</sub> pH  | 7.49(0.03)          | 7.56(0.11)         |
| Bicarbonate-extractable P (mmol kg <sup>-1</sup> )          | 0.17(0.03)          | 0.10(0.03)         |
| Soil-solution Ca <sup>2+</sup> (mmol L <sup>-1</sup> )      | 2.41(0.34)          | 1.19(0.22)         |
| Soil-solution $K^+$ (mmol $L^{-1}$ )                        | 0.10(0.01)          | 0.15(0.05)         |
| Soil-solution $SO_4^{2-}$ (mmol L <sup>-1</sup> )           | 0.19(0.03)          | 0.20(0.12)         |
| Extractable K <sup>+</sup> (mmol kg <sup>-1</sup> )         | 0.60(0.04)          | 0.26(0.03)         |
| Extractable Ca <sup>2+</sup> (mmol kg <sup>-1</sup> )       | 18.02(1.40)         | 10.47(0.46)        |
| Extractable Mg <sup>2+</sup> (mmol kg <sup>-1</sup> )       | 0.53(0.04)          | 0.50(0.13)         |

<sup>&</sup>lt;sup>a</sup>Data of average of 6 subsamples of stock soil. Standard deviations in parentheses. See methods section for references on procedures.

Table 2. Plant growth responses to atmospheric carbon dioxide enrichment for high and low fertility soils<sup>a</sup>

|                              | Н          | igh fertility soil |                  | Low fertility soil |            |          |  |  |
|------------------------------|------------|--------------------|------------------|--------------------|------------|----------|--|--|
| Attribute                    | Ambient    | Elevated           | % change         | Ambient            | Elevated   | % change |  |  |
| Above-ground mass (g)        | 7.31(1.31) | 9.80(1.21)         | +34              | 0.78(0.25)         | 1.22(0.50) | +56      |  |  |
| Leaf area (cm <sup>2</sup> ) | 735(137)   | 740(134)           | +0.7             | 79(24)             | 85(28)     | +8       |  |  |
| Number of leaves             | 13.2(2.9)  | 13.1(2.6)          | -0.7             | 5.3(1.9)           | 6.0(1.8)   | +13      |  |  |
|                              |            | ANOVA              | probability valu | ies                |            |          |  |  |
|                              | Above-gre  | ound mass          | Leaf area        | # le               | aves       |          |  |  |
| CO <sub>2</sub>              | < 0.001    |                    | 0.830            | 0.805              |            |          |  |  |
| Soil                         | <0         | .001               | < 0.001          | 0.0                |            |          |  |  |
| $CO_2 \times Soil$           | 0          | .001               | 0.979            | 0.6                | 541        |          |  |  |

<sup>&</sup>lt;sup>a</sup>Standard deviations in parentheses. Above-ground mass measured after drying at 60 °C for 48 h.

bagged along with measured leaves. Soil in individual pots was homogenized and a subsample reserved. All samples were shipped overnight from Texas on dry ice to the Reno, NV USDA-Agricultural Research Service soil and plant analysis laboratory and kept in a 4  $^{\circ}$ C refrigerator until processed ( $\leq$  14 days).

Above-ground tissue was dried at 60 °C for 48 h, weighed and ground in a commercial mill. Total C, N and S were quantified on subsamples with a CHNS analyzer. Another subsample was ashed at 500 °C, solubilized in 1N HCl and analyzed for P (molybdenum blue chemistry), Ca, Mg, Fe, Mn, Cu, Zn (atomic adsorption spectroscopy), and Na and K (atomic emission spectroscopy) (Kalra, 1998). Roots were frozen,

then a known weight was blended with a known weight of deionized ice for 2 min. A portion of the slurry was analyzed for amidase, urease, asparaginase, and glutaminase enzyme activities (Tabatabai, 1994). Another portion of the root slurry was filtered and the filtrate was analyzed for ortho-P, citrate, and malate by ion chromatography. Soil was analyzed on a fresh weight basis and recalculated to 105 °C weight on a separate subsample. Enzyme activities of acid phosphatase, amidase, urease, asparaginase, and glutaminase were measured using standard methods (Tabatabai, 1994). Available N was extracted with 2 N KCl (Bundy and Meisinger, 1994). A 30 day aerobic incubation procedure was used as a proxy for N min-

*Table 3.* Above-ground tissue nutrient concentration and selected tissue elemental mole ratios in response to atmospheric CO<sub>2</sub> enrichment by soil fertility level<sup>a</sup>

| Attribute                        |                | Ambient  | _     | h fertility s<br>Elevated |       | Change      | Ambi      |         | ow fertilit<br>Elevated | •     | % Change |
|----------------------------------|----------------|----------|-------|---------------------------|-------|-------------|-----------|---------|-------------------------|-------|----------|
| C (mmol g <sup>-1</sup>          | )              | 33.4(0.4 | 4)    | 33.3(0.8)                 | _     | -0.3        | 30.80     | (1.5)   | 31.9(1.                 | 6)    | +3.6     |
| N (mmol g <sup>-1</sup>          | )              | 1.04(0   | 0.20) | 0.74(0.0                  | 8) –2 | 28.8        | 1.4       | 1(0.32) | 0.99(                   | 0.25) | -29.8    |
| S ( $\mu$ mol g <sup>-1</sup> )  | )              | 50.3(6.0 | 0)    | 45.7(3.3)                 | _     | -9.1        | 91.80     | (16.4)  | 79.5(1                  | 7.2)  | -13.4    |
| $P(\mu \text{mol } g^{-1})$      | )              | 33.8(4.2 | 2)    | 27.5(8.6)                 | -1    | 18.6        | 66.6      | (20.2)  | 55.1(25                 | 5.5)  | -17.3    |
| Ca (mmol g                       | <sup>1</sup> ) | 0.72(0   | 0.12) | 0.58(0.1                  | 1) -1 | 19.4        | 0.90      | 6(0.20) | 0.92(                   | 0.20) | -4.2     |
| Mg (μmol g                       | ·1)            | 125(14)  |       | 101(11)                   | -1    | 19.2        | 181(3     | 5)      | 145(20)                 |       | -19.9    |
| $K (\mu \text{mol g}^{-1})$      | )              | 61(7)    |       | 48(3)                     | -2    | 21.3        | 56(1      | 0)      | 49(13)                  |       | -12.5    |
| Fe ( $\mu$ mol g <sup>-1</sup>   | <sup>1</sup> ) | 3.7(1.   | 7)    | 3.2(0.9)                  | -1    | 13.5        | 8.9       | (5.0)   | 5.1(1.                  | 7)    | -42.7    |
| Mn (μmol g                       | ·1)            | 2.13(0   | 0.35) | 2.06(0.4                  | 2) –  | -3.3        | 2.29      | 90.39)  | 2.02(0                  | 0.36) | -11.8    |
| C:N                              |                | 32.9(4.9 | 9)    | 45.4(5.8)                 | +3    | 38.3        | 22.90     | (5.6)   | 34.6(10                 | 0.2)  | +51.1    |
| N:S                              |                | 20.8(3.4 | 4)    | 16.4(2.3)                 | -2    | 21.2        | 15.90     | (4.6)   | 12.6(2.                 | 3)    | -20.8    |
|                                  |                |          |       |                           | ANOV  | A probabili | ty values |         |                         |       |          |
|                                  | C              | N        | S     | P                         | Ca    | Mg          | K         | Fe      | Mn                      | C:N   | N:S      |
| $CO_2$                           | 0.694          | 0.161    | 0.367 | 0.598                     | 0.261 | 0.055       | 0.233     | 0.480   | 0.344                   | 0.208 | 0.039    |
| Soil                             | 0.085          | 0.085    | 0.029 | 0.061                     | 0.011 | < 0.001     | 0.750     | 0.190   | 0.570                   | 0.047 | 0.022    |
| $\text{CO}_2 \times \text{Soil}$ | 0.456          | 0.658    | 0.648 | 0.765                     | 0.506 | 0.367       | 0.645     | 0.458   | 0.355                   | 0.797 | 0.604    |

<sup>&</sup>lt;sup>a</sup>Standard deviation in parentheses.

eralization potential (Hart et al., 1994) with net N mineralization determined by subtracting KCl-extractable N. Quantification of NO<sub>3</sub> and NO<sub>2</sub> used ion chromatography after removal of Cl<sup>-</sup> with a colloidal silver filter. Quantification of NH<sub>4</sub><sup>+</sup> used a membrane diffusion colorimetric procedure. Cations were extracted by ammonium acetate (Thomas, 1982) with Ca<sup>2+</sup> and Mg<sup>2+</sup> quantified by atomic adsorption spectroscopy and Na<sup>+</sup> and K<sup>+</sup> quantified by atomic emission spectroscopy. Cations and anions in the soil-solution were extracted using immiscible displacement with CCl<sub>4</sub> (Mubarek and Olson, 1976). Anions in the soilsolution were quantified by ion chromatography and atomic adsorption/emission spectroscopy was used to quantify Ca<sup>2+</sup>, Mg<sup>2+</sup>, Na<sup>+</sup>, and K<sup>+</sup>. We measured the bicarbonate-extractable pool of available P (Olsen and Sommers, 1982). For all analytical measurements only certified standards were used which were made up in the same matrixes of the various extraction methods.

Data were analyzed using a mixed model splitplot ANOVA (SAS, 1999), with glasshouse bay (2 bays per CO<sub>2</sub> treatment) as the blocking factor. The experimental design was 2 replications of 6 (plant treatments) and 3 (controls) subsamples for each soil fertility level and CO<sub>2</sub> level. Analysis of plant attributes used categorical variables CO<sub>2</sub> (ambient and elevated) and Soil (high fertility and low fertility soils) with glasshouse bay( $CO_2$ ) the error term for  $CO_2$  and soil  $\times$  glasshouse bay( $CO_2$ ) the error term for soil and the  $CO_2 \times Soil$  interaction. Analysis of soil attributes used categorical variables  $CO_2$ , Soil, and Plant (planted with L. latifolium or unplanted controls) with glasshouse bay( $CO_2$ ) the error term for  $CO_2$  and soil  $\times$  plant  $\times$  bay( $CO_2$ ) the error term for soil, plant and the soil  $\times$  plant interaction. Confidence intervals at the 90% level were generated to compare means.

#### Results

#### Plant attributes

Above-ground biomass of *L. latifolium* was influenced by a significant (P < 0.001)  $CO_2 \times$  soil interaction (Table 2). Biomass increased with  $CO_2$  enrichment by 34% in the high fertility soil and by 56% in the low fertility soil. Growth in the high fertility soil produced significantly ( $P \le 0.01$ ) greater number of leaves and nearly 10 times greater leaf area than growth in the low fertility soil, but neither differed between  $CO_2$  treatments.

Overall, above-ground tissue elemental concentration declined in the elevated  $CO_2$  treatment; however, only Mg was significant (P = 0.055) (Table 3). Soil

Table 4. Biomass produced per unit nutrient in response to atmospheric CO<sub>2</sub> treatments for high and low fertility soils<sup>a</sup>

| Nutrient         | Ambient    | High fertilit<br>Elevated | y soil | % Increase | Ambient           | Low ferti<br>Elevat | •       | % Inc | rease |
|------------------|------------|---------------------------|--------|------------|-------------------|---------------------|---------|-------|-------|
| С                | 2.49(0.03) | 2.50(                     | 0.06)  | 0.4        | 2.71(0.13)        | ) 2.6               | 1(0.13) | 3.7   |       |
| N                | 69.7(10.4) | 97.1(1                    | 0.7)   | 39.3       | 52.8(11.4)        | 76.50               | (19.7)  | 44.9  |       |
| S                | 627(72)    | 686(50)                   |        | 9.4        | 349(62)           | 409(8               | 8)      | 17.2  |       |
| P                | 974(134)   | 1287(430                  | ))     | 32.1       | 554(198)          | 720(3               | 42)     | 30.0  |       |
| Ca               | 35.4(5.9)  | 43.9(7                    | .5)    | 24.0       | 27.2(6.0)         | 28.60               | (8.1)   | 5.1   |       |
| Mg               | 331(43)    | 413(42)                   |        | 24.8       | 238(41)           | 289(4               | 1)      | 21.4  |       |
| K                | 43.2(5.0)  | 53.9(3                    | .7)    | 24.8       | 47.6(11.2)        | 55.80               | (13.0)  | 17.2  |       |
| Zn               | 25.0(3.4)  | 27.0(4                    | .9)    | 8.0        | 27.6(4.1)         | 35.90               | (15.3)  | 30.1  |       |
|                  |            |                           |        | ANOVA p    | probability value | s                   |         |       |       |
|                  | С          | N                         | S      | P          | Ca                | Mg                  | K       | Z     | Zn .  |
| $CO_2$           | 0.678      | 0.176                     | 0.254  | 4 0.51     | 14 0.207          | 0.067               | 0.276   | 0     | .527  |
| Soil             | 0.105      | 0.055                     | 0.003  | < 0.00     | 0.025             | 0.015               | 0.592   | 0     | .419  |
| $CO_2 \times So$ | oil 0.502  | 0.674                     | 0.97   | 7 0.24     | 17 0.327          | 0.324               | 0.824   | 0     | .611  |

<sup>&</sup>lt;sup>a</sup>Standard deviations in parentheses. All units are g of biomass produced per g of nutrient, except for Zn which is kg of biomass per g.

fertility level was a main effect explaining plant elemental concentration; except for C, plants grown in the low fertility soil had significantly (P < 0.10) higher concentrations of N, S, P, Ca, and Mg compared to plants grown in the high fertility soil. Carbon to N ratios of above-ground tissue from plants increased with  $CO_2$  enrichment for both soils, but this effect was not statistically significant. The ratio of N:S declined significantly (P = 0.039) with  $CO_2$  enrichment.

Biomass produced per unit of nutrient (also referred in the literature as nutrient use efficiencies) increased with  $CO_2$  enrichment but only Mg exhibited a significant (P=0.067) effect (Table 4). The high fertility soil had significantly ( $p \leq 0.055$ ) greater biomass production per unit of N, S, P, Ca, and Mg higher than the low fertility soil.

Concentrations of ortho-P, citrate, malate and enzyme activities of amidase and asparaginase in roots of *L. latifolium* were similar between  $CO_2$  treatments (Table 5). Of these attributes, only concentration of ortho-P was significantly (P = 0.044) affected by soil type being higher in root growth in the low fertility soil.

#### Soil Attributes

As expected, the high fertility soil had greater concentrations or activities of most soil attributes measured than the low fertility soil (Table 6). Carbon dioxide enrichment did affect availability of some nutrients.

Acetate extractable Mg<sup>2+</sup> was influenced by a significant (P = 0.063)  $CO_2 \times soil$  interaction;  $CO_2$ enrichment increased Mg<sup>2+</sup> in the high fertility soil, but decreased Mg<sup>2+</sup> in the low fertility soil. Available N was influenced by a significant (P = 0.056) CO<sub>2</sub> × soil × plant interaction; CO<sub>2</sub> enrichment induced greater available N in the planted than control pots in the high fertility soil only. The bicarbonate-extractable pool of available P was significantly (P = 0.050)greater in soil exposed to CO<sub>2</sub> enrichment regardless of whether the pots were planted or not. Concentration of soil-solution  $SO_4^{2-}$  was affected by a significant  $(P = 0.042) \text{ CO}_2 \times \text{soil interaction}; \text{CO}_2 \text{ enrichment}$ increased  $SO_4^{2-}$  availability in planted and control high fertility soil. A significant (P = 0.045) CO<sub>2</sub> × soil interaction affected concentration of soil-solution  $Mg^{2+}$ ;  $CO_2$  enrichment reduced available  $Mg^{2+}$  in both planted and control pots of the low, but not high, fertility soil. Enzyme activities of amidase (P =0.252), glutaminase (P = 0.966), and phosphatase (P = 0.302) were not significantly affected by  $CO_2$ treatment.

# Discussion

Carbon dioxide enrichment influenced many plant and soil responses of the invasive crucifer *L. latifolium* in both low and high fertility soils, but did not influence amidohydrolase activities nor acid phosphatase activ-

Table 5. Concentration of ortho-P, citrate, malate and amidase and asparaginase activities in roots in response to atmospheric CO<sub>2</sub> enrichment by soil fertility level<sup>a</sup>

|   |         | High fer    | tility soil      |         | Low ferti    | lity soil   |
|---|---------|-------------|------------------|---------|--------------|-------------|
| Attribute   |         | Ambient     | Elevated         |         | Ambient      | Elevated    |
| Ortho-P ( $\mu$ mol g <sup>-1</sup> )                       |         | 9.07(1.98)  | 8.99(3.43)       |         | 19.74(12.85) | 18.89(5.75) |
| Citrate ( $\mu$ mol g <sup>-1</sup> )                       |         | 11.99(0.71) | 11.91(0.59)      |         | 16.35(2.88)  | 11.84(0.79) |
| Malate ( $\mu$ mol g <sup>-1</sup> )                        |         | 8.02(2.58)  | 8.43(2.09)       |         | 12.66(10.96) | 8.56(2.67)  |
| Amidase ( $\mu$ mol g <sup>-1</sup> hr <sup>-1</sup> )      |         | 10.75(3.29) | 8.02(2.40)       |         | 18.19(17.11) | 10.07(6.98) |
| Asparaginase ( $\mu$ mol g <sup>-1</sup> hr <sup>-1</sup> ) |         | 1.09(0.33)  | 1.05(0.27)       |         | 2.42(2.01)   | 1.63(1.54)  |
|   |         | ANOV        | A probability va | alues   |              |             |
|   | Ortho-P | Citrate     | Malate           | Amidase | Asparaginase |             |
| $CO_2$  | 0.921   | 0.354       | 0.531            | 0.232   | 0.279        |             |
| Soil  | 0.044   | 0.170       | 0.166            | 0.145   | 0.127        |             |
| $CO_2 \times Soil$  | 0.806   | 0.156       | 0.190            | 0.270   | 0.295        |             |

<sup>&</sup>lt;sup>a</sup>Standard deviations in parentheses. All values based on fresh root weight.

ity; thus, hypothesis 1 is rejected. We were unable to find literature citations documenting atmospheric CO<sub>2</sub> relationships with soil amidohydrolase activities. The literature does; however, suggest that CO<sub>2</sub> enrichment can significantly increase soil phosphatase activity (Barrett et al., 1998; Kang et al., 2001; Moorhead and Linkins, 1997). Possibilities that might explain the lack of a phosphatase-CO2 effect in our study include: (1) the experiment may have been of insufficient time to witness an effect; (2) pot studies may under-evaluate such CO<sub>2</sub> effects; and (3) low organic P availability relative to inorganic P availability in the two soils used. The lack of an amidohydrolase-CO2 effect is more perplexing because the high fertility soil had significantly more available N upon plant harvest with CO<sub>2</sub> enrichment than at ambient CO<sub>2</sub>. Given that this increase in available N must occur through organic matter mineralization, one would expect the activities of some amidohydrolases to be higher with CO<sub>2</sub> enrichment to cleave amide groups into the plant available NH<sub>4</sub><sup>+</sup> form. It is possible that our conservative statistical design is overlooking what are actually statistically significant differences in amidohydrolase activities in response to CO2 enrichment. Alternatively, N mineralizing enzymes not measured in this study may actually control the kinetics of organic matter mineralization. In either case, the fact that available N in the high fertility soil increased in both planted and unplanted controls suggests that CO<sub>2</sub> enrichment influences the soil microbial community regardless of an interaction with plant root exudation.

Hypothesis 2 is rejected because CO<sub>2</sub> enrichment did not increase citrate and malate in roots of *L. latifo*-

lium. One explanation for this is that the soils used in this study, which represent the range of soil fertility levels being invaded, are not fully representative of soils L. latifolium is invading; many invaded soils are saline and/or sodic, contain calcium carbonate, and in general have high nutrient availability (Blank and Young, 2002). We discount this possibility because citrate and malate have been shown to be effective in releasing P bound to Al and Fe mineral surfaces characteristic of soils used in this pot study (Penaloza et al., 2002; Shen et al., 2002). In addition, CO<sub>2</sub> enrichment does not necessarily increase root exudation (Niklaus et al., 2001), even exudation of citrate (Barrett and Gifford, 1999). If this is true for L. latifolium, there may be no benefit to this plant in producing higher concentrations of citrate and malate in roots with CO<sub>2</sub> enrichment if it is not to be exuded. Fine root turnover has been shown to increase with CO<sub>2</sub> enrichment (Luo et al., 2001), but given the short-term nature of this study, it is impossible to say if fine root turnover over long time periods with CO<sub>2</sub> enrichment might indeed increase plant contributions of acetate and malate to the soil, and thereby increase P availability.

Carbon dioxide enrichment did not significantly increase pools of available soil Ca which suggests rejection of hypothesis 3. *Lepidium latifolium* uptakes considerable Ca and has been shown to increase levels of soil-solution Ca relative to the grass *Elytrigia elong-ata* it is competing with, presumably to meet nutritional needs (Blank and Young, 2002). In addition, soil pools of Ca have previously been demonstrated to be affected by CO<sub>2</sub> enrichment (Hagedorn et al., 2002). It is possible that hypothesis rejection is comprom-

Table 6. Selected soil attributes following conclusion of experiment in control and planted pots by soil fertility and carbon dioxide level<sup>a</sup>

| $ \begin{array}{ l c c c c c c c c c c c c c c c c c c $  |   |                           |           |                       | High fertility soil | ity soil    |          |              |             | Low fertility soil | lity soil          |            | Ì                       |         |
|---|---|---------------------------|-----------|-----------------------|---------------------|-------------|----------|--------------|-------------|--------------------|--------------------|------------|-------------------------|---------|
| Ambient         Elevated         1.39(0.7)         3.4(0.3)         0.30(0.3)         0  |   |                           | +T        | latifoliur            | и                   | Coi         | ntrol    | <u> </u><br> | + L. lati,  | folium             | Cor                | ıtrol      | ĺ                       |         |
| 13.9(0.7)         14.8(4.0)         12.8(1.4)         15.9(4.2)         9.1(2.0)         8.3(0.5)         8.4(0.3)         10.0(4.4)           0.75(0.03)         0.83(0.24)         0.75(0.11)         0.91(0.24)         0.30(0.05)         0.27(0.03)         0.30(0.03)         0.38(0.21)           0.69(0.06)         0.75(0.20)         0.61(0.04)         0.70(0.19)         0.57(0.12)         0.44(0.06)         0.64(0.14)         0.55(0.24)           1.72(0.80)         1.65(0.68)         1.21(0.83)         1.28(0.32)         0.13(0.08)         0.01(0.06)         0.07(0.02)         0.064(0.14)         0.55(0.24)           1.20(0.65)         1.74(0.56)         1.28(0.40)         2.09(0.67)         0.07(0.02)         0.07(0.02)         0.10(0.04)         0.07(0.02)         0.10(0.04)         0.07(0.02)         0.10(0.04)         0.10(0.02)         0.10(0.04)         <   | Attribute                                   |                           | Ambient   | Eleva                 | ]<br>               | Ambient     | Elevated | <br> -       |             | Elevated           | Ambient            | Elevated   | I                       |         |
| 0.95(0.03)         0.83(0.24)         0.75(0.11)         0.91(0.24)         0.30(0.05)         0.20(0.03)         0.30(0.03)         0.30(0.03)         0.30(0.03)         0.30(0.03)         0.30(0.03)         0.30(0.03)         0.30(0.03)         0.30(0.04)         0.55(0.24)         0.55(0.24)         0.55(0.24)         0.55(0.24)         0.55(0.24)         0.55(0.24)         0.55(0.24)         0.55(0.24)         0.01(0.06)         0.01(0  | Ext. $Ca^{2+}$ (mmol kg <sup>-1</sup>       | (-)                       | 13.9(0.7) |                       | 3(4.0)              | 12.8(1.4)   | 15.9(4.2 |              | .1(2.0)     | 8.3(0.5)           | 8.4(0.3)           | 10.0(4.4)  | I                       |         |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$  | Ext. $K^+$ (mmol kg <sup>-1</sup> )         |                           | 0.75(0.0  | _                     | 33(0.24)            | 0.75(0.11)  | 0.91(0   | _            | .30(0.05)   | 0.27(0.03)         | 0.30(0.03)         | 0.38(0.21  | _                       |         |
| 1.72(0.80)         1.65(0.68)         1.21(0.83)         1.28(0.32)         0.13(0.08)         0.05(0.08)         0.01(0.06)         0.07(0.02)         0.07(0  | Ext. $Mg^{2+}$ (mmol kg <sup>-</sup>        | .1)                       | 0.69(0.0  | _                     | 75(0.20)            | 0.61(0.04)  | 0.70(0   | _            | .57(0.12)   | 0.44(0.06)         | 0.64(0.14)         | 0.55(0.24  |                         |         |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | Net N min. (mmol kg-                        | .1)                       | 1.72(0.8  | _                     | 65(0.68)            | 1.21(0.83)  | 1.28(0   | _            | 13(0.08)    | 0.05(0.08)         | 0.01(0.06)         | 0.07(0.08  |                         |         |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$  | KCl N (mmol kg <sup>-1</sup> )              |                           | 1.20(0.0  | _                     | 74(0.56)            | 1.89(0.40)  | 2.09(0   | _            | 0.07(0.02)  | 0.07(0.02)         | 0.08(0.03)         | 0.10(0.01  |                         |         |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | Bicarb-P (mmol kg <sup>-1</sup> )           |                           | 0.21(0.0  | _                     | (30.06)             | 0.23(0.04)  | 0.28(0   | _            | .08(0.02)   | 0.08(0.02)         | 0.07(0.02)         | 0.10(0.04  | _                       |         |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | Amidase ( $\mu$ mol g <sup>-1</sup> hr      | ·1)                       | 14.16(1.4 | _                     | 58(4.30)            | 15.18(1.74) | 17.60(5  | _            | 1.85(0.36)  | 0.71(0.10)         | 0.79(0.14)         | 0.82(0.41  |                         |         |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$  | Glutaminase ( $\mu$ mol g <sup>-</sup>      | $^{-1}   hr^{-1})$        | 15.60(5.9 |                       | 7(4.36)             | 17.12(4.57) | 20.14(3  |              | (98.0)89    | 0.62(0.55)         | 1.10(0.76)         | 0.46(0.43  |                         |         |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$  | Phosphatase (µmol g <sup>-</sup>            | $^{1}  \mathrm{hr}^{-1})$ | 2.65(0.4  | _                     | 78(0.94)            | 2.92(0.40)  | 3.76(1   |              | .45(0.20)   | 0.41(0.10)         | 0.33(0.26)         | 0.41(0.19) | _                       |         |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$   | ID $SO_4^{2-}$ (mmol L <sup>-1</sup> )      |                           | 5.3(3.4)  |                       | (3.2)               | 10.1(2.4)   | 13.4(10  |              | (6.0)9:     | 3.7(2.4)           | 13.2(4.6)          | 7.9(1.6)   |                         |         |
| 3.26(0.49)         3.71(0.76)         3.15(0.83)         7.27(4.08)         7.24(3.00)         5.52(2.07)         7.47(4.22)         3.35(1.76)           1.23(0.32)         1.05(0.39)         1.29(0.30)         1.32(0.30)         0.56(0.28)         0.48(0.04)         0.52(0.07)         0.50(0.02)           Ext. K <sup>+</sup> Ext. Mg <sup>2+</sup> Net N min.         KC1 N         Bicarb-P         Amidase         Glutaminase.         Phosphatase         ID SO <sub>4</sub> <sup>2</sup> D.50(0.02)           0.173         0.703         0.588         0.074         0.050         0.252         0.966         0.302         0.871         0.769           0.338         0.063         0.831         0.116         0.367         0.145         0.228         0.677         0.042         0.549           0.302         0.783         0.639         0.836         0.109         0.109         0.513         0.213         0.677         0.042         0.549           0.306         0.783         0.639         0.836         0.084         0.265         0.892         0.935         0.250         0.248         0.548           0.906         0.999         0.691         0.994         0.852         0.589         0.952         0.951         0.951   | ID $\text{Ca}^{2+}$ (mmol L <sup>-1</sup> ) |                           | 105(14)   | 115(2                 | (7;                 | 128(38)     | 118(23)  | 2            |             | 53(15)             | 77(33)             | 65(23)     |                         |         |
| L1.23(0.32)         1.05(0.39)         1.29(0.30)         1.32(0.30)         0.56(0.28)         0.48(0.04)         0.52(0.07)         0.50(0.02)           Ext. $K^+$ Ext. $M_g^{2+}$ Net N min.         KCl N         Bicarb-P         Amidase Glutaminase.         Phosphatase ID $SO_2^4$ ID $Ca^{2+}$ 0.173         0.703         0.588         0.074         0.050         0.252         0.966         0.302         0.871         0.769           0.001         0.013         <0.001  | ${ m ID}{ m Mg}^{2+}({ m mmol}{ m L}^{-1})$ |                           | 3.26(0.   | _                     | 71(0.76)            | 3.15(0.83)  | 7.27(4   |              | .24(3.00)   | 5.52(2.07)         | 7.47(4.22)         | 3.35(1.76  |                         |         |
|   | Total C $(\text{mol kg}^{-1})$              |                           | 1.23(0.3  | _                     | (60.36)             | 1.29(0.30)  | 1.32(0   |              | .56(0.28)   | 0.48(0.04)         | 0.52(0.07)         | 0.50(0.02  |                         |         |
|   |   |                           |           |                       |                     |             | ANOV     | A probabi    | lity values |                    |                    |            |                         |         |
|   | 田   | xt. Ca <sup>2+</sup>      |           | 3xt. Mg <sup>2+</sup> |                     | KCIN        | Bicarb-P | Amidase      | Glutamina   |                    | ase ID $SO_4^{2-}$ |            | $^{10}\mathrm{Mg}^{2+}$ | Total C |
| Soil <a.0.001< th=""> <th< td=""><td></td><td>0.161</td><td></td><td>.703</td><td>0.588</td><td>0.074</td><td>0.050</td><td>0.252</td><td>996.0</td><td>0.302</td><td>0.871</td><td>0.769</td><td>0.852</td><td>0.378</td></th<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<></a.0.001<> |   | 0.161                     |           | .703                  | 0.588               | 0.074       | 0.050    | 0.252        | 996.0       | 0.302              | 0.871              | 0.769      | 0.852                   | 0.378   |
| 0.328 0.338 0.063 0.831 0.116 0.367 0.145 0.228 0.677 0.042 0.549 0.549 0.745 0.302 0.783 0.120 0.119 0.100 0.513 0.281 0.670 <0.001 0.196 0.196 0.182 0.348 0.638 0.836 0.084 0.265 0.892 0.935 0.250 0.296 0.548 0.800 0.976 0.103 0.291 0.291 0.590 0.482 0.985 0.985 0.997 0.991 0.998 0.999 0.691 0.056 0.904 0.852 0.588 0.952 0.071 0.593  |   | :0.001                    |           | 0.013                 | <0.001              | <0.001      | < 0.001  | <0.001       | < 0.001     | < 0.001            | 0.065              | 0.001      | 0.190                   | <0.001  |
| 0.745 0.302 0.783 0.120 0.119 0.100 0.513 0.281 0.670 <0.001 0.196 0.108 0.182 0.348 0.638 0.836 0.084 0.265 0.892 0.935 0.250 0.296 0.548 0.080 0.976 0.103 0.291 0.213 0.231 0.590 0.482 0.085 0.085 0.937 0.997 0.991 0.999 0.691 0.056 0.904 0.852 0.588 0.952 0.071 0.593 0.993 0.993 0.993 0.993 0.994 0.995 0.994 0.995 0.994 0.995 0.994 0.995 0.994 0.995 0.994 0.995 0.994 0.995 0.994 0.995 0.994 0.995 0.995 0.994 0.995 0.994 0.995  |   | 0.328                     |           | .063                  | 0.831               | 0.116       | 0.367    | 0.145        | 0.228       | 0.677              | 0.042              | 0.549      | 0.045                   | 0.843   |
| 0.182 0.348 0.638 0.836 0.084 0.265 0.892 0.935 0.250 0.296 0.548 0.548 0.800 0.976 0.103 0.291 0.213 0.231 0.590 0.482 0.085 0.432 0.997 0.91ant 0.938 0.906 0.999 0.691 0.056 0.904 0.852 0.588 0.952 0.071 0.593 0.  |   | 0.745                     |           | .783                  | 0.120               | 0.119       | 0.100    | 0.513        | 0.281       | 0.670              | < 0.001            | 0.196      | 0.725                   | 0.246   |
| 0.800 0.976 0.103 0.291 0.213 0.231 0.590 0.482 0.085 0.432 0.997 0.991 0.938 0.906 0.999 0.691 0.056 0.904 0.852 0.588 0.952 0.071 0.593   |   | 0.182                     |           | .638                  | 0.836               | 0.084       | 0.265    | 0.892        | 0.935       | 0.250              | 0.296              | 0.548      | 0.790                   | 0.330   |
| 0.938 0.906 0.999 0.691 0.056 0.904 0.852 0.588 0.952 0.071 0.593   | Soil $\times$ Plant                         | 0.800                     |           | ).103                 | 0.291               | 0.213       | 0.231    | 0.590        | 0.482       | 0.085              | 0.432              | 0.997      | 0.229                   | 0.193   |
|   | $CO_2 \times Soil \times Plant$             | 0.938                     | 0.906     | 666'                  | 0.691               | 0.056       | 0.904    | 0.852        | 0.588       | 0.952              | 0.071              | 0.593      | 0.192                   | 0.635   |

 $^{4}$ Standard deviations are provided in parentheses. Abbreviations are as follows: Ext. = pH 7.0 ammonium acetate extractable, Bicarb-P = bicarbonate extractable P, ID = immiscibly displaced which is equivalent to the soil-solution pool.

ised by a conservative statistical design. Indeed, with  $CO_2$  enrichment there was an increase in the acetate-extractable pool of  $Ca^{2+}$  and greater plant uptake of  $Ca^{2+}$ . Moreover, proper testing of this hypothesis may have required a soil more characteristic of where *L. latifolium* is invading presently.

Hypothesis 4 addressed the influence of CO<sub>2</sub> enrichment on plant-soil relationships of L. latifolium to investigate the possibility that CO<sub>2</sub> enrichment is allowing L. latifolium to become more competitive. Because our experimental design did not incorporate interspecific competition, a direct assessment of the influence of CO<sub>2</sub> enrichment on competitiveness of L. latifolium is not possible. In a study of the invasive species Centaurea solstitialis, Dukes (2002) determined that for individual plants, growth was greatly enhanced due to CO<sub>2</sub> enrichment, but in competition with representative grass species, the competitive advantage incurred was relatively small. As a generalization, CO2 growth enhancement measured on individual plants will not proxy for those same plants grown in interspecific competition (Poorter and Navas, 2002). Moreover, CO<sub>2</sub> stimulated growth enhancement may be a poor predictor of competitive success compared to other attributes such as increases in cover (Stewart and Potvin, 1996) and the ability to fix N (Poorter and Navas, 2002). Nonetheless, this study provides evidence that CO<sub>2</sub> enrichment interacts with the plant and soil in particular ways which suggests the possibility of strengthened competitiveness. Carbon dioxide enrichment significantly increased the available pool of soil N (high fertility soil only) and increased the bicarbonate pool of extractable P. Previous field and greenhouse experiments suggest that N and P availability are critical in explaining success of L. latifolium (Blank et al., 2002; Blank and Young, 2002). Such increases in availability of these nutrients, especially in nutrient poor environments, may be sufficient to tip the competitive advantage in favor of L. latifolium. Carbon dioxide enrichment provides another potential competitive advantage to L. latifolium by allowing greater accumulation of plant biomass per unit of nutrient (nutrient use efficiency), which is reported in the literature for many plant species, with some exceptions (Baxter et al., 1994; Davey et al., 1999; Fangmeier et al., 1997; Hagedorn et al., 2002; Johnson et al., 1995). Any competitive benefit L. latifolium receives would of course be mitigated by the relative increase of biomass to nutrient uptake of competing plants. Even so, a plant that can produce greater biomass using less nutrients and can increase competitive attributes such as shading ability, seed production, rhizome formation etc. would, as a first approximation, be more competitive. A confounding factor in judging the increased competitive ability of *L. latifolium* to CO<sub>2</sub> enrichment is determining how increasing atmospheric CO<sub>2</sub> has affected, and may continue to affect, the plant-soil relationships of *L. latifolium* along the continuum from pre-industrial to predicted future levels of CO<sub>2</sub>. For example, some weedy species have gained greater competitive benefits with increases in CO<sub>2</sub> from pre-industrial to present-day concentrations compared to benefits that are expected to result in future CO<sub>2</sub> enriched environments (Ziska, 2003).

### Acknowledgements

Authors wish to thank Ms Tye Morgan, Ms Lisa Prinz, Mr Kyle Tiner and Mr Chris Kolodziejczyk for assistance in setting up and maintaining experiment and laboratory analyses.

#### References

- Barrett D J, Richardson A E and Gifford R M 1998 Elevated atmospheric CO<sub>2</sub> concentrations increase wheat root phosphatase activity when growth is limited by phosphorus. Aust. J. Plant Physiol. 25, 87–93.
- Barrett D J and Gifford R M 1999 Increased C-gain by an endemic Australian pasture grass at elevated atmospheric  $CO_2$  concentration when supplied with non-labile inorganic phosphorus. Aust. J. Plant Physiol. 26, 443–451.
- Baxter R, Gantley M, Ashenden T W, and Farrar J F 1994 Effects of elevated carbon dioxide on three grass species from montane pasture II. Nutrient uptake, allocation and efficiency of use. J. Exp. Bot. 45, 1267–1278.
- Bazin A, Goverde M, Erhardt A, and Shykoff J A 2002 Influence of atmospheric carbon dioxide enrichment on induced response and growth compensation after herbivore damage in *Lotus corniculatus*. Ecol. Entomol. 27, 271–278.
- Bazzaz F A and Garbutt K 1988 The response of annuals in competitive neighborhoods: Effect of elevated CO<sub>2</sub>. Ecol. 69, 937–946.
- Blank R R, Qualls R G, and Young J A 2002 Lepidium latifolium: Plant nutrient competition-soil interactions. Biol. Fert. Soils 35, 458–464.
- Blank R R 2002 Amidohydrolase activity, soil N status, and the invasive crucifer *Lepidium latifolium*. Plant Soil 239, 155–163.
- Blank R R and Young J A 2002 Influence of the exotic invasive crucifer, *Lepidium latifolium*, on soil properties and elemental cycling. Soil Sci. 167, 821–829.
- Bundy L G and Meisinger J J 1994 Nitrogen availability indices. In Methods of Soil Analysis, Part 2 Microbiological and Biochemical Properties. Ed. R W Weaver et al. pp. 951–984. Soil Sci. Soc. Amer. Inc., Madison, WI.
- Campbell C D and Sage R F 2002 Interactions between atmospheric CO<sub>2</sub> concentration and phosphorus nutrition on the formation

- of proteoid roots in white lupin (*Lupinus albus* L.) Plant Cell Environ. 25, 1051–1059.
- Davey P A, Parson A J, Atkinson L, Wadge K and Long S P 1999 Does photosynthetic acclimation to elevated CO<sub>2</sub> increase photosynthetic nitrogen-use efficiency? A study of three native UK grassland species in open-top chambers. Funct. Ecol. 13, 21–28.
- Derner J D, Johnson H B, Kimball B A, Pinter Jr. P J, Polley H W, Tischler C R, Boutton T W, LaMorte R L, Wall G W, Adam N R, Leavitt S W, Ottman M J, Matthias A D and Brooks T J 2002 Above- and belowground responses of C<sub>3</sub>-C<sub>4</sub> species mixtures to elevated CO<sub>2</sub> and soil water availability. Global Change Biol. 9, 452–460.
- Dukes J S and Mooney H A 1999 Does global change increase the success of biological invaders? Trends Ecol. Evol. 14, 135–139.
- Dukes J S 2002 Comparison of the effect of elevated CO<sub>2</sub> on an invasive species (*Centaurea solstitialis*) in monoculture and community settings. Plant Ecol. 225, 225–234.
- Fangmeier A, Grüters U, Högy P, Vermehren B and Jäger H J 1997 Effect of elevated CO<sub>2</sub>, nitrogen supply and tropospheric ozone on spring wheat – II. Nutrients (N, P, K, S, Ca, Mg, Fe, Mn, Zn). Environ. Poll. 96, 43–59.
- Hagedorn F, Landolt W, Tarjan D, Egli P, and Bucher J B 2002 Elevated CO<sub>2</sub> influences nutrient availability in young beech-spruce communities on two soil types. Oecologia 132, 109–117.
- Hart S C, Stark J M, Davidson E A and Firestone M K 1994 Nitrogen mineralization, immobilization, and nitrification. *In Methods of Soil analysis part 2 Microbiological and Biochemical Properties*. Ed. R W Weaver et al. pp. 985–1018. Soil Sci. Soc. Amer. Inc., Madison, WI.
- Hoffland, E, Van den Boogaard, R, Nelemans J, and Findenegg G 1992 Biosynthesis and root exudation of citric and malic acids in phosphate-starved rape plants. New Phytol. 122, 675–680.
- Johnson D W, Ball T and Walker R F 1995 Effects of elevated carbon dioxide and nitrogen on nutrient uptake in ponderosa pine seedlings. Plant Soil 168–169, 535–545.
- Jones C G, Lawton J H, and Schachak M 1994 Organisms as ecosystem engineers. Oikos 69, 373–386.
- Kalra Y P 1998 Handbook of reference methods for plant analysis. CRC Press, Boca Raton FL, 300 pp.
- Kang H, Freeman C, and Ashendon T W 2001 Effects of elevated CO<sub>2</sub> on fen peat biogeochemistry. Sci. Total Environ. 279, 45–50
- Luo Y, Wu L, Andrews J A, White L, Matamala R, Schafer K V R, and Schlesinger W H 2001 Elevated CO<sub>2</sub> differentiates ecosystem carbon processes: Deconvolution analysis of Duke Forest face data. Ecol. Monogr. 71, 357–376.

- Marks S M and Strain B 1989 Effects of drought and  $CO_2$  enrichment on competition between two old-field perennials. New Phytol. 111, 181–186.
- Moorhead D L and Linkins A E 1997 Elevated CO<sub>2</sub> alters belowground exoenzyme activities in tussock tundra. Plant Soil 189, 321–329.
- Mubarek A and Olsen R A 1976 Immiscible displacement of the soil solution by centrifugation. Soil Sci. Soc. Amer. J. 40, 329–331.
- Niklaus P A, Glockler E, Siegwolf R, and Korner C 2001 Carbon allocation in calcareous grassland under elevated CO<sub>2</sub>: A combined <sup>13</sup>C pulse-labelling/soil physical fractionation study. Funct. Ecol. 15, 43–50.
- Olsen S R and Sommers L E 1982 Phosphorus In Methods of Soil Analysis Part 2 Chemical and Microbiological Properties. Ed. A L Page. pp. 403–430. Am. Soc. Agron. Inc., Madison, WI.
- Penaloza E, Corcuera L J, and Martinez J 2002 Spatial and temporal variation in citrate and malate exudation and tissue concentration as affected by P stress in roots of white lupin. Plant Soil 241, 200–221
- Poorter H and Navas M 2002 Plant growth and competition at elevated CO<sub>2</sub>: on winners, losers and functional groups. New Phytol. 157, 175–198.
- SAS Institute. 1999 SAS System. Version 8. SAS, Cary, NC.
- Shen H, Yan-X, Zhao M, Zheng S, and Wang X 2002 Exudation of organic acids in common bean as related to mobilization of aluminum- and iron-bound phosphates. Environ. Exp. Bot. 48, 1–9.
- Stewart J and Potvin C 1996 Effect of elevated CO<sub>2</sub> on an artificial grassland community: competition, invasion and neighbourhood growth. Funct. Ecol. 10, 157–166.
- Tabatabai M A 1994 Soil enzymes. In Methods of Soil Analysis Part 2 Microbiological and Biochemical Properties. Ed. R W Weaver et al. Soil Sci. Soc. Amer. Inc., Madison, WI.
- Thomas G W 1982 Cation exchange capacity. In Methods of Soil Analysis Part 2 Chemical and Microbiological Properties. Ed. A L Page et al. Soil Sci. Soc. Amer. Inc., Madison WI.
- Young J A, Turner C E, and James L F 1995 Perennial pepperweed. Rangelands 17, 121–123.
- Ziska L H 2001 Change in competitive ability between a C<sub>4</sub> crop and a C<sub>3</sub> weed with elevated carbon dioxide. Weed Sci. 49, 622–627.
- Ziska L H 2003 Evaluation of the growth response of six invasive species to past, present and future atmospheric carbon dioxide. J. Exper. Bot. 54, 395–404.

Section editor: H. Lambers