

## United States Department of Agriculture- Agricultural Research Service-Greenhouse Production Research Group

Modified Hoagland's Standard Nutrient Recipe

Stock Solution Recipes		
	Concentration of	
Compound	stock solution	
KH <sub>2</sub> PO <sub>4</sub>	1 <i>M</i> = 136.09 g/L	
KNO <sub>3</sub>	1 <i>M</i> = 101.11 g/L	
$Ca(NO_3)_2$	1 <i>M</i> = 236.15 g/L	
MgSO <sub>4</sub>	1 <i>M</i> = 246.48 g/L	
Fe-DTPA	0.071 <i>M</i> = 33.24 g/L	
MnCl <sub>2</sub>	0.01 <i>M</i> = 1.98 g/L	
CuCl <sub>2</sub>	0.01 <i>M</i> =1.7 g/L	
ZnCl <sub>2</sub>	0.01 <i>M</i> = 1.3 g/L	
H <sub>3</sub> BO <sub>3</sub>	0.1 <i>M</i> = 6.18 g/L	
MoNa <sub>2</sub> O <sub>4</sub>	0.001 <i>M</i> = 0.242 g/L	

Final nutrient solution should be adjusted to pH between 5.6 and 6.0

	Volume of stock per liter of final solution	Element	Final concentration of element
	mL		
Macronutrients			тM
KH <sub>2</sub> PO <sub>4</sub>	2.0	Р	2.0
		К	4.5*
KNO <sub>3</sub>	2.5	Ν	7.5**
Ca(NO <sub>3</sub> ) <sub>2</sub>	2.5	Ca	2.5
MgSO₄	1.0	Mg	1.0
		S	1.0
Micronutrients			μ <i>Μ</i>
Fe-DTPA	1.0	Fe	71.0
MnCl <sub>2</sub>	0.9	Mn	9.0
CuCl <sub>2</sub>	0.15	Cu	1.5
ZnCl <sub>2</sub>	0.15	Zn	1.5
H₃BO₃	0.45	В	45.0
MoNa <sub>2</sub> O <sub>4</sub>	0.1	Mo	0.1

\*K concentration calculated from KH<sub>2</sub>PO<sub>4</sub> and KNO<sub>3</sub>

\*\*N concentration calculated from KNO<sub>3</sub> and Ca(NO<sub>3</sub>)<sub>2</sub>